# Preparation of Immobilized Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub> Catalyst and Its Catalytic performance in Hydroformylation of Mixed Octenes

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### Introduction

The practical industrial applications of homogeneous catalysis have been often limited by the difficulties of the separation of catalysts and products [1,2]. The entrapment of homogeneous catalysts inside the porous silica matrices prepared by the sol-gel method appears to be a promising strategy for catalyst recovery [3]. Although the entrapment of hydroformylation catalysts inside the porous inorganic matrices has been reported [4], there are no studies concerning the hydroformylation of internal and branched long chain olefins (such as, mixed octenes) with hybrid matrices as support of immobilized catalysts. In addition, the triphenylphosphine oxide (Ph<sub>3</sub>PO), which is regarded as a weak ligand, exhibited a positive effect on the hydroformylation of mixed octenes to C<sub>9</sub> aldehydes in a homogeneous reaction catalyzed by Rh catalysts [5]. In this paper, we report catalytic efficiency of immobilized Co-Ph<sub>3</sub>PO supported on modified SiO<sub>2</sub> with polydimethylsiloxane (PDMS) as the catalysts for hydroformylation of mixed octenes with a good recycle ability.

#### Materials and Methods

Co(CH<sub>3</sub>COO)<sub>2</sub>...4H<sub>2</sub>O, polydimethylsiloxane (PDMS), tetraethylorthosilicate (TEOS) and Ph<sub>3</sub>PO were used for preparing immobilized Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub> catalyst. Certain amount of Co(CH<sub>3</sub>COO)<sub>2</sub>·4H<sub>2</sub>O and Ph<sub>3</sub>PO were dissolved in THF solvent in a Schlenk flask. While stirring the above solution, deionized water, TEOS, methanol and PDMS were added to the solution. The resulted mixture was allowed for gelation. The formed gel was washed with  $\mathrm{CH}_2\mathrm{Cl}_2$ and dried at 353 K, and the Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub> catalyst was obtained [4]. The hydroformylation of the mixed octenes (containing n-octenes, methyl heptenes, dimethyl hexenes and trimethyl pentanes) was carried out in an autoclave (80ml). The products were analyzed by a gas chromatograph.

## **Results and Discussion**

Figure 1 shows the FT-IR spectra of the immobilized catalysts with PDMS modification and without PDMS (Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub> and Co-Ph<sub>3</sub>PO/SiO<sub>2</sub>). The bands observed at 3058, 1439 cm<sup>-1</sup> were assigned to C-H symmetric stretching of phenyl and C-C asymmetric stretching of the phenyl ring, respectively [6]. The absorption peaks at 2967, 2892, 1264, 1000-1100, 850, 800 cm<sup>-1</sup> were from PDMS [7,8]. The peaks at 2967 and 2892 cm<sup>-1</sup> were assigned to the symmetric and asymmetric C-H stretching vibration of methyl groups, and strong doublet peak at 1050 to 1100 cm<sup>-1</sup> and 800 cm<sup>-1</sup> were due to Si-O-Si bonds. A strong absorption peak at 1264 cm<sup>-1</sup> was the characteristic of methyl groups bonded to silicon and attributed the symmetric deformation of C-H [7,8]. The above IR spectroscopic results suggest that the significant interaction occurred between the OH groups at the surface of SiO<sub>2</sub> and PDMS.

The immobilized catalyst Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub> could be easily separated from the reaction mixtures and reused for several times, and showed an obvious advantage in recycle use over the homogeneous catalyst. The recycling of the catalysts, Co-Ph<sub>3</sub>PO/SiO<sub>2</sub> and Co-Ph<sub>3</sub>PO/PDMS /SiO2, was performed through a series of consecutive runs in the hydroformylation of the mixed octenes. The results are presented in Figure 2. The immobilized catalyst without PDMS modification (Co-Ph<sub>3</sub>PO/SiO<sub>2</sub>), deactivated quickly. However, in the case of the immobilized catalyst with PDMS modification (Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub>), the results showed clearly that the catalyst could be recycled at least six times with only a slight loss in activity, with the exception of the catalytic activity increasing in the first recycle.

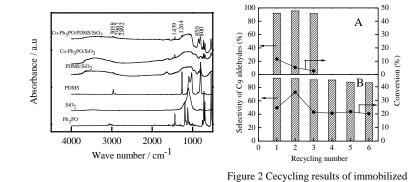


Figure 1 FT-IR spectra for several samples

catalysts in hydroformylation of mixed octenes (A)Co-Ph<sub>3</sub>PO/SiO<sub>2</sub>, (B)Co-Ph<sub>3</sub>PO/PDMS/SiO<sub>2</sub>.

## Significance

The hydroformylation of mixed octeness is an important process for producing valuable isononyl aldehyde, which is one of the raw materials of di-iso-nony phthalate (a high performance plasticizer). Molecular catalysts immobilized on solid supports have advantages for catalyst recycle. The results obtained in this work implied that the organic-inorganic hybrid support (PDMS-SiO2) had positive effect on the recycle of the immobilized catalyst in the hydroformylation of mixed octenes.

#### References

- Herrmann, W.A., Cornils, B. Angew. Chem., Intl. Ed. Engl. 36, 1048 (1997). 1.
- Bailey, D.C., Langer, S.H. Chem. Rev. 81, 109 (1981). 2.
- 3. Davis, M.E. Chemtech 29, 32 (1999).
- Campos, de J.D.R., Buffon, R. New J. Chem., 27, 446 (2003). 4.
- 5. He, D.H., Pang, D.C., Chen, Y., et al. Journal of Molecular Catalysis A 174, 21(2001)
- Westermark, G., Persson, I. Colloids and Surfaces A 144, 149 (1998). 6.
- Shindou, T., Katayama, S., Yamada, N., et al. J. Sol-Gel Sci. and Tech. 30, 229 (2004). 7.
- 8.
- Julian, B., Cervais, C., Cordoncillo, E., et al. Chem. Mater., 15, 3026 (2003).